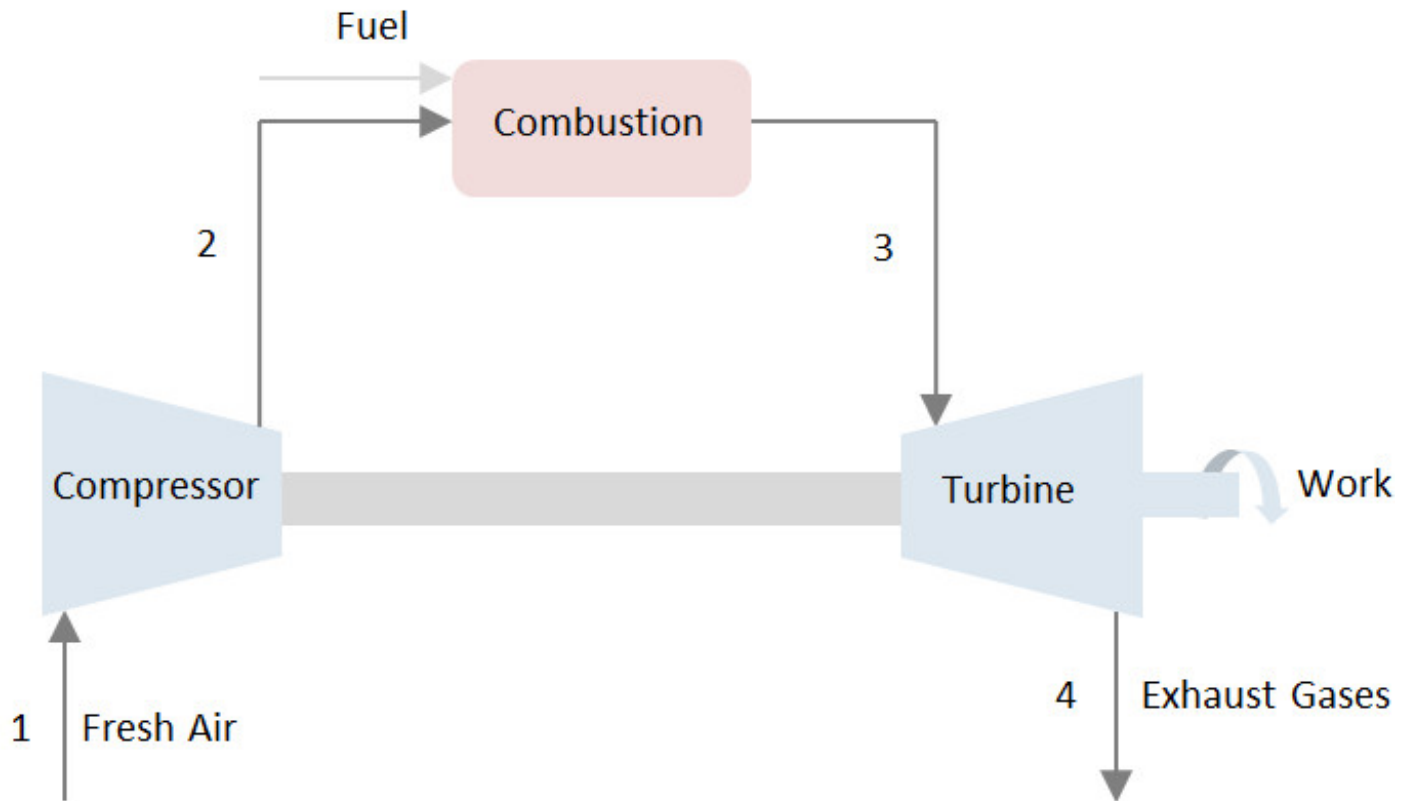


Ideal Brayton Cycle

▼ Introduction

Consider this ideal Brayton cycle.



The compressor inlet temperature and pressure is 300 °C and 101 kPa, the compression ratio is 8, and the combustor outlet temperature is 1300 °C. The compressor and turbine are isentropic, and combustion is isobaric.

What is the thermal efficiency of the system?

▼ Parameters

```
> with(ThermophysicalData) :  
  with(Units[Standard]) :
```

Compressor Inlet Temperature

```
> Tmin := 300degC :
```

Compressor Exit Temperature

```
> Tmax := 1300degC :
```

Compressor inlet pressure

> $P_{\min} := 101\text{kPa}$:

Compressor pressure ratio

> $r := 8$:

▼ Compressor Inlet

> $P_1 := P_{\min}$:

> $T_1 := T_{\min}$:

> $h_1 := \text{Property}(\text{enthalpy, temperature} = T_1, \text{pressure} = P_1, \text{air})$

$$h_1 := 7.052729614 \cdot 10^5 \frac{\text{J}}{\text{kg}}$$

> $s_1 := \text{Property}(\text{entropy, temperature} = T_1, \text{pressure} = P_1, \text{air})$

$$s_1 := 4547.480132 \frac{\text{J}}{\text{kg K}}$$

▼ Combustor Inlet

> $s_2 := s_1$:

> $P_2 := r \cdot P_1$:

> $T_2 := \text{Property}(\text{temperature, pressure} = P_2, \text{entropy} = s_2, \text{air})$

$$T_2 := 991.5856803 \text{ K}$$

> $h_2 := \text{Property}(\text{enthalpy, temperature} = T_2, \text{pressure} = P_2, \text{air})$

$$h_2 := 1.163321097 \cdot 10^6 \frac{\text{J}}{\text{kg}}$$

▼ Turbine Inlet

> $P_3 := P_2$:

> $T_3 := T_{\max}$:

> $h_3 := \text{Property}(\text{enthalpy, temperature} = T_3, \text{pressure} = P_3, \text{air})$;

$$h_3 := 1.851847268 \cdot 10^6 \frac{\text{J}}{\text{kg}}$$

> $s_3 := \text{Property}(\text{entropy, } T = T_3, P = P_3, \text{air})$;

$$s_3 := 5092.509817 \frac{\text{J}}{\text{kg K}}$$

▼ Turbine Exit

> $P_4 := P_1$:

> $s_4 := s_3$:

> $T_4 := \text{Property}(\text{temperature, pressure} = P_4, \text{entropy} = s_4, \text{air})$

$$T_4 := 947.0939895 \text{ K}$$

> $h_4 := \text{Property}(\text{enthalpy, temperature} = T_4, \text{pressure} = P_4, \text{air})$;

$$h4 := 1.112422426 \cdot 10^6 \frac{\text{J}}{\text{kg}}$$

▼ Thermal Efficiency

$$> \eta := 1 - \frac{h4 - h1}{h3 - h2}$$

$$\eta := 0.4086652304$$